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# UTILITY PATENT APPLICATION TRANSMITTAL

Attorr	ney Docket No.	06076 USA
		Joel E. Goldstein
		dehyde Nonwoven Binders Which .

Express Mail Label No. EL685717250US (Only for new nonprovisional applications under 37 CFR 1.53(b)) Assistant Commissioner for Patents APPLICATION ELEMENTS ADDRESS TO: **Box Patent Application** Washington, DC 20231 See MPEP chapter 600 concerning utility patent application contents Fee Transmittal Form (e.g., PTO/SB/17) CD-ROM or CD-R in duplicate, large table or (Submit an original and a duplicate for fee processing) Computer Program (Appendix) Applicant claims small entity status. 8. Nucleotide and/or Amino Acid Sequence Submission See 37 CFR 1.27. (if applicable, all necessary) pecification [Total Pages 12] referred arrangement set forth below) Specification X Computer Readable Form (CRF) - Descriptive title of the invention Specification Sequence Listing on: Cross Reference to Related Applications i. ☐ CD-ROM or CD-R (2 copies); or - Statement Regarding Fed sponsored R & D - Reference to sequence listing, a table, ii. D paper or a computer program listing appendix Statements verifying identity of above copies - Background of the Invention Brief Summary of the Invention
Brief Description of the Drawings (if filed) **ACCOMPANYING APPLICATION PARTS** - Detailed Description Assignment Papers (cover sheet & document(s)) - Claim(s) Power of 37 CFR 3.73(b) Statement - Abstract of the Disclosure 10. (when there is an assignee) Attorney English Translation Document (if applicable) Drawing(s) (35 U.S.C. 113) [ Total Sheets [ ] Copies of IDS Information Disclosure [Total Pages 3] 5. Oath or Declaration Citations Statement (IDS)/PTO-1449 Newly executed (original or copy)
Copy from a prior application (37 CFR 1.63 (d))
(for continuationIdivisional with Box 17 completed) Preliminary Amendment Return Receipt Postcard (MPEP 503) (Should be specifically itemized) **DELETION OF INVENTOR(S)** Certified Copy of Priority Document(s) (if foreign priority is claimed) 15 Signed statement attached deleting inventor(s) named in the prior application, see 37 CFR 16. 1.63(d)(2) and 1.33(b). Application Data Sheet. See 37 CFR 1.76 17. If a CONTINUING APPLICATION, check appropriate box, and supply the requisite information below and in a preliminary amendment, or in an Application Data Sheet under 37 CFR 1.76: of prior application No.: Continuation Divisional Continuation-in-part (CIP) Prior application information: Group I Art Unit-For CONTINUATION OR DIVISIONAL APPS only: The entire disclosure of the prior application, from which an oath or declaration is supplied under Box 5b, is considered a part of the disclosure of the accompanying continuation or divisional application and is hereby incorporated by reference. The incorporation <u>can only</u> be relied upon when a portion has been inadvertently omitted from the submitted application parts. 18. CORRESPONDENCE ADDRESS Customer Number or Bar Code Label Correspondence address below Name 23543 Air Products and Chemicals, Inc. **Address** City State Zip Code Telephone Country Fax 25,073 Russéll L. Brewer Registration No. (Attorney/Agent) Name (Print/Type) 10/30/2000 Signature Dung Date

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			First Named Inventor	Joel E. Goldstein
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METHOD OF PAYMENT	FEE CALCULATION (continued)	
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Check Credit card Money Order Other	113 1,840* 113 1,840* Requesting publication of SIR after Examiner action	
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1. BASIC FILING FEE	116 390 216 195 Extension for reply within second month	
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114 150 214 75 Provisional filing fee	138 1,510 138 1,510 Petition to institute a public use proceeding	
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<b>SUBTOTAL (1)</b> (\$) 710	141 1,240 241 620 Petition to revive - unintentional	
2. EXTRA CLAIM FEES	142 1,240 242 620 Utility issue fee (or reissue)	
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Independent	122 130 122 130 Petitions to the Commissioner	
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SUBMITTED BY			Complete (if applicable)			
Name (Print/Type)	Russell L. Brewer	Registration No. (Attorney/Agent)	25,073	Telephone	610-481-7289	
Signature	Lessell Dres			Dat <del>e</del>	10/30/2000	

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#### TITLE OF THE INVENTION:

## REDUCED FORMALDEHYDE NONWOVEN BINDERS WHICH CONTAIN POLYMERIZED UNITS OF N-METHYLOLACRYLAMIDE

#### BACKGROUND OF THE INVENTION

Nonwoven webs which are comprised of loosely assembled webs or masses of fibers bound together with an adhesive binder have many applications. These include paper towels, disposable diapers, filtration products, disposable wipes, and the like. Vinyl acetate based emulsions incorporating crosslinkable functionality are widely used in the preparation of these nonwoven products. One of the favorite crosslinking systems is based upon N-methylolacrylamide.

The emission of formaldehyde represents a worker safety and health issue, as well as a consumer issue, and there have been significant attempts to prepare adhesive binders having either reduced formaldehyde content or they are formaldehyde-free. Reduction of formaldehyde in vinyl acetate based emulsions has been achieved by using less favored reducing agents to the formaldehyde sulfoxylates, or by reducing the level of N-methylol acrylamide employed in the adhesive binder. Formaldehyde-free binders eliminate both the N-methylol acrylamide and the use of any formaldehyde emitting reducing agent. However, either method for producing vinyl acetate based emulsion polymers for nonwoven webs has been done so at the expense of performance and cost.

Representative patents disclosing adhesive binders having a post-curable comonomer such as N-methylolacrylamide are as follows:

U.S. 3,081,197 discloses a nonwoven binder incorporating an internal plasticizer and a post-curable comonomer such as N-methylolacrylamide.

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U.S. 3,380,851 discloses a binder comprising an interpolymer of vinyl acetate/ethylene/N-methylolacrylamide for the preparation of nonwoven goods.

U.S. 4,449,978 discloses vinyl acetate/ethylene/N-methylolacrylamide/acrylamide systems for producing nonwoven webs. The advantage of using the N-methylolacrylamide/acrylamide blend has been the ability to reduce the amount of formaldehyde present in the emulsion and in the cured web without a substantial loss of performance.

U.S. 5,540,987 discloses vinyl acetate/ethylene/ N-methylolacrylamide polymers for use in producing nonwoven webs having reduced formaldehyde content as well as reduced formaldehyde content in the cured web. Reduced formaldehyde content is achieved by using a redox system based upon a hydrophobic hydroperoxide and ascorbic acid. These systems result in formaldehyde contents significantly lower than those systems produced using a corresponding reducing agent incorporating no formaldehyde such as sodium metabisulfite or formaldehyde emitting reducing agents such as sodium formaldehyde sulfoxylate.

Several of the approaches to reduced formaldehyde content in nonwoven webs have been directed to actually prepare formaldehyde free nonwoven binders. The approaches pursued for the former include the use of crosslinkers based upon acrylamidobutyraldehyde dialkyl acetal, the methacrylamide derivative, the cyclized version, methyl acrylamidoglycolate methyl ether, allyl glycidyl ether, 3-chloro-2-hydroxypropyl (meth)acrylate and acrylamidoglycolic acid. The problems with these kinds of crosslinkers for nonwoven binders included poorer performance (lower tensile development), activation of the crosslinking chemistry at a pH which is corrosive to the commercial machinery, the requirement to use a more expensive and commercially unavailable novel self-crosslinking monomer and the instability of the self-crosslinking monomer.

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#### BRIEF SUMMARY OF THE INVENTION

The invention relates to improved vinyl acetate based emulsion polymers incorporating N-methylolacrylamide as a crosslinking component for construction of nonwoven webs. The emulsions polymers are prepared by emulsion polymerization of vinyl acetate, N-methylolacrylamide and optional monomers in the presence of a stabilizing system and a redox catalyst system comprised of an oxidizing agent and a reducing agent. The improvement for achieving reduced formaldehyde in the emulsion polymer resides in the use of a glycolic acid adduct of sodium sulfite (a proprietary reducing agent sold under the trademark Bruggolite FF-6) as the reducing agent. Lower formaldehyde levels are achieved in contrast to other formaldehyde free reducing agents such as sodium erythorbate and sodium ascorbate, and to sodium formaldehyde sulfoxylate, or any of the other reducing agents typically used in the preparation of a nonwoven binder employing N-methylolacrylamide as the self-crosslinking monomer.

There are significant advantages associated with the use of the glycolic acid adduct of sodium sulfite in producing vinyl acetate based emulsion binders containing N-methylolacrylamide. They include:

an ability to generate vinyl acetate-based emulsions which are highly suited for use in preparing nonwoven products having substantially reduced formaldehyde levels in the emulsion or latex; and,

an ability to generate nonwoven products having excellent resistance to solvents and water while at the same time having excellent adhesion to the fibers for providing enhanced tensile strength.

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The N-methylolacrylamide containing emulsions for use in producing the nonwoven webs are based on copolymers of vinyl acetate and N-methylolacrylamide. Typically, these are vinyl acetate/ethylene/ N-methylolacrylamide containing polymers, although optionally other monomers may be included in the polymer system as desired. The ethylene content will range broadly from about 10 to 40% and preferably 15 to 30% by weight, yielding an adhesive having a Tg of from -15 to 10°C. Other monomers which may be included in the adhesive binder include vinyl esters of aliphatic carboxylic acids as well as C<sub>1</sub>-C<sub>8</sub> alkyl acrylates and methyacrylates. Examples include methylacrylate, methyl methacrylate, ethyl acrylate, butyl acrylate, and 2-ethylhexylacrylate. Small amounts of polyolefinically-unsaturated monomers, e.g., dioctyl and dibutyl maleate, and the like may also be used.

Commercially, N-methylolacrylamide used in the preparation of vinyl acetate-based nonwoven binders is obtained with about a 48% aqueous solution with up to about 2% acrylamide. Another type of N-methylolacrylamide product which may be used herein is an N-methylolacrylamide/acrylamide mixture. Often this blend is present in a 1:1 molar ratio. Isobutoxy methyl acrylamide is an optional crosslinker but is not preferred to N-methylolacrylamide.

Monomers other than vinyl acetate and ethylene are polymerized in amounts of less than 10% by weight and generally even less than 5% by weight. These levels are exclusive of the level of N-methylolacrylamide or of the level of N-methylolacrylamide blend used in preparing the emulsion polymer. The crosslinking monomer N-methylolacrylamide is generally incorporated at levels of from 0.5 to 10% by weight, although typically levels are from about 1.5 to 5% by weight of the polymer.

Stabilizing systems used for preparing the vinyl acetate based polymers include nonionic emulsifiers such as polyoxyethylene condensates of the formula  $R-(CH_2CH_2O)_n-H$  wherein R is the residue of fatty alcohol containing 10 to 18 carbon

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atoms, an alkyl phenol, a fatty acid containing 10 to 18 carbon atoms, an amide or amine, and the like. Examples of polyethoxylated condensates are based on ethoxylated tridecyl alcohol, fatty alcohols, i.e., lauryl alcohol, octyl phenol, nonyl phenol or a secondary alcohol. Furthermore, the stabilizing package may include sodium vinyl sulfonate, sodium dodecyl benzene sulfonate, dioctyl sulfosuccinate, disodium lauryl sulfosuccinate, sodium lauryl sulfate, sodium methyl 2-sulfolaurate and the sodium or ammonium salts of the sulfonates or phosphates of any of the above described nonionic surfactants. Often combinations of emulsifying agents are used; e.g., a relatively hydrophobic emulsifying agent used in combination with the relatively hydrophilic agent.

The amount of active surfactants in the stabilizing system based upon total monomers typically ranges from 1.5% to 6.0%, preferably from 2.5% - 4.0% by weight. The ratio between the individual components in the surfactant stabilizing package is one of design.

One of the keys to producing the emulsion polymers having reduced free formaldehyde in the latex as well as in the dried cured web is in the initiator system used to effect polymerization of the monomers. The initiator system employed herein is a redox system based upon an oxidizing agent and a particular class of reducing agents. A wide variety of oxidizing agents may be used. Preferably, these oxidizing agents are inorganic and hydrophobic peroxides, such as hydrogen peroxide, *t*-butylhydroperoxide and benzoyl peroxide. Thermal initiators such as the persulfates, e.g., ammonium and potassium persulfate, may be used.

Although nonformaldehyde-containing reducing agents, e.g., sodium metabisulfite and ascorbic acid or alkali metal salt thereof have been used in the past in combination with an oxidizing agent, it has been found that a class of sulfinic acid derivatives and particularly the glycolic acid adduct of sodium sulfite affords an

exceptional reduction in free formaldehyde content in the emulsions as compared to the other nonformaldehyde emitting reducing agents.

The sulfinic acid derivatives are represented by the formula:

$$\begin{array}{c|c} O & R_1 \\ \hline S & R_2 \\ \hline R_3 \end{array}$$

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where M is a hydrogen atom an ammonium atom or a monovalent metal ion, e.g., sodium, potassium or an equivalent of a divalent metal, e.g., zinc;  $R_1$  is OH or  $NR_4R_5$ wherein R<sub>4</sub> and R<sub>5</sub> each are H or C<sub>1</sub>-C<sub>6</sub> alkyl; R<sub>2</sub> is H or an alkyl, alkenyl, cycloalkyl or aryl and the like, and R<sub>3</sub> is CO<sub>2</sub>M. The preferred reducing agent is formed by reacting sodium dithionite with glycolic acid. Other examples of sulfinic acid compounds include 2-hydroxyphenyl hydroxymethyl sulfinic acid-sodium salt; 4-methoxyphenyl hydroxymethyl sulfinic acid-sodium salt; 2-hydroxy-2-sulfinato acetic acid-disodium salt; 2-hydroxy-2-sulfinato acetic acid-zinc salt; 2-hydroxy-2-sulfinato propionatic acid-disodium salt; ethyl 2-hydroxy-2-sulfinato propionate-sodium salt, and the like.

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This class of reducing agents is the only one that seems to offer these significant reductions in vinyl acetate-based adhesives for use in preparing nonwoven products utilizing N-methylolacrylamide as the crosslinking monomer. Preferred results are obtained with the glycolic acid adduct of sodium sulfite which is sold under the trademark Bruggolite FF-6.

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The initiator system comprised of an oxidizing and reducing agent is employed in an amount of from about 0.1 to 2% by weight of the total monomers utilized in the polymerization process. The molar ratio of oxidizing agent to reducing agent typically is greater than 1. Conventionally the molar amount of the reducing agent used is less than the molar amount of the oxidizing agent.

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Temperatures for effecting polymerization range from about 25 to 85°C with typical polymerizations being operated in the range of from 55 to 70°C. Reactor pressures range from about 300 to 1200 psig and are dependent upon the level of ethylene that is desired to be incorporated into the polymer.

The manufacture of nonwoven webs is well known, and exemplary processes are noted in U.S. 4,449,978 which is incorporated by reference. Conventionally a mass of fibers are deposited or arranged on a web by an air-laid, etc., technique and then contacted with an aqueous polymeric binder in amounts sufficient to provide about 10 to 100%, by weight of the emulsion polymer, on a dry basis, based on the weight of the starting web. The water is removed and the polymer crosslinked through its self-crosslinking mechanism by use of an acid catalyst. Drying temperatures typically range from 150 to 200°F for about 4 to 6 minutes followed by a cure of 300 to 310°F for 3 to 5 minutes.

A wide variety of fibrous materials may be used for preparing nonwoven webs, including cellulosic fibers, polyester, polyolefin, polyurethane, etc. Cellulose is one of the more common fibrous materials which is used for producing paper nonwovens such as disposable diapers, consumer towels, disposable wipes, and filtration products.

The following examples are provided to illustrate various embodiments of the invention and are not intended to restrict the scope thereof.

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#### Example 1

Polymerization of Vinyl Acetate/Ethylene/N-methylolacrylamide Binder
Using Glycolic Acid Adduct of Sodium Sulfate Reducing Agent

The polymerization of various vinyl acetate-ethylene/N-methylolacrylamide systems was carried out in a one gallon stirred, stainless steel reaction vessel equipped with a jacket. The reaction vessel was charged initially with 800.0 g of deionized water,

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305.0 g of sodium octylphenol ethoxylate sulfate, 3EO (sold under the trademark Polystep C-OP3S), 0.91 g of sodium citrate, 2.4 g of 50% aqueous citric acid, 2.3 g of 5% aqueous ferric ammonium sulfate and 1558.0 g of vinyl acetate. While stirring, 240.0 g of ethylene was introduced below the surface of the liquid in the reaction vessel in order that the interpolymers would have a vinyl acetate:ethylene weight ratio of about 80:20. The reaction vessel was heated to 50 °C.

The following three aqueous solutions were intermittently added to the reaction vessel over the course of the reaction (on a delay basis); (a) 7.5% Bruggolite FF-6, (b) 3.0 % t-butylhydroperoxide and (c) 291.6 g of a 48% aqueous solution of functional monomers (a commercially available material comprised of approximately 28% aqueous **NMA** 20% and aqueous acrylamide). After four hours the Nmethylolacrylamide/acrylamide (MAMD) delay was complete and the other two delays continued for another 30 minutes. The reaction was terminated by cooling.

The resulting dispersion contained 51.5% solids, a pH of 5.3, and a Brookfield viscosity of 182 cps at 60 rpm and a #3 spindle. The glass transition temperature,  $T_g$ , of the polymer was 8.9 °C. The grit level of the dispersion was 156 ppm on a 100 mesh screen and 25 ppm on a 325 mesh screen.

#### Example 2

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Comparison of Vinyl Acetate/Ethylene/N-methylolacrylamide Binders

Using Glyolic Acid Adduct of Sodium Sulfate Reducing Agent and Ascorbic Acid

Three comparative vinyl acetate/ethylene/self-crosslinking monomer dispersions were prepared where the only significant differences to the Example 1 formulation was in

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the amount of the reducing agent used in the redox couple and in the amount of ethylene in the resultant polymer.

The first vinyl acetate/ethylene/N-methylolacrylamide emulsions polymers were low  $T_g$  dispersions, -14 °C. The dispersion formed with the Bruggolite FF-6 based system gave a formaldehyde level of 3.3 ppm while the sodium erythorbate based system gave a formaldehyde level of 27.2 ppm.

The second vinyl acetate/ethylene/N-methylolacrylamide emulsion polymers were high  $T_g$  dispersions, 10 °C. The Bruggolite FF-6 based systems gave a formaldehyde level of 8.6 ppm while the sodium erythorbate based systems gave a formaldehyde level of 57.1 ppm.

The third second vinyl acetate/ethylene/N-methylolacrylamide emulsion polymers were medium  $T_g$  dispersions, 0 °C, but the level of self-crosslinking monomer was 50% higher than either of the other two examples. The Bruggolite FF-6 based system gave a formaldehyde level of 6.8 ppm while the sodium erythorbate based system gave a formaldehyde level of 47.5 ppm.

In summary, these data show that not only does the Bruggolite FF6 reducing agent based the presumed glycolic acid adduct of sodium sulfite provide excellent results in terms of polymerization rates but is also more effective in reducing the free formaldehyde in the emulsion than other formaldehyde free reducing agent, ascorbic acid. Lower free formaldehyde content in the emulsion may also result in al lower level of free formaldehyde level in the final dried and cured nonwoven web. Further, the tensile properties and absorbencies of the webs produced using the Bruggolite FF-6 reducing agent are indistinguishable one using the ascorbic acid as the reducing agent, all other parameters remaining the same.

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#### **CLAIMS**

What is claimed is:

1. In a vinyl acetate based polymer based emulsion formed by the polymerization of vinyl acetate and N-methylolacrylamide, optionally other monomers, in the presence of a stabilizing system and a redox catalyst system comprised of an oxidizing agent and a reducing agent, the improvement for reducing formaldehyde emissions in the emulsion, which comprises:

forming said vinyl acetate based polymer emulsion utilizing as the reducing component of the redox catalyst system a reducing agent of the formula:

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$$\begin{array}{c} O \\ S \\ \hline \\ R_3 \end{array}$$

where M is a hydrogen atom, an ammonium atom or a monovalent metal ion,  $R_1$  is OH or  $NR_4R_5$  wherein  $R_4$  and  $R_5$  each are H or  $C_1$ - $C_6$  alkyl;  $R_2$  is H or an alkyl, alkenyl, cycloalkyl or aryl and the like, and  $R_3$  is  $CO_2M$ .

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2. The vinyl acetate polymer based emulsion of Claim 1 in which the vinyl acetate polymer-based emulsion comprises ethylene in an amount of from about 10 to 40% by weight of the polymer.

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3. The vinyl acetate polymer based emulsion of Claim 2 wherein the N-methylolacrylamide is present in an amount of from about 0.5 to 10% by weight of the polymer.

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- 4. The vinyl acetate polymer based emulsion of Claim 3 wherein the reducing agent represented by the formula is selected from the group consisting of: 2-hydroxyphenyl hydroxymethyl sulfinic acid-sodium salt; 4-methoxyphenyl hydroxymethyl sulfinic acid-sodium salt; 2-hydroxy-2-sulfinato acetic acid-disodium salt; 2-hydroxy-2-sulfinato acetic acid-zinc salt; 2-hydroxy-2-sulfinato propionatic acid-disodium salt; ethyl 2-hydroxy-2-sulfinato propionate-sodium salt.
- 5. The vinyl acetate polymer based emulsion of Claim 4 wherein the vinyl acetate-based emulsion polymer is formed using a redox catalytic system of hydrophobic hydroperoxide and the glycolic acid adduct of sodium sulfonate.
- 6. The vinyl acetate polymer based emulsion of Claim 3 wherein M is sodium or zinc.
- The vinyl acetate polymer based emulsion of Claim 3 wherein R₁ is OH.

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#### ABSTRACT OF THE DISCLOSURE

The invention is the use of a glycolic acid adduct of sodium sulfite as the reducing agent in the redox couple for polymerization of vinyl acetate polymer based emulsions for nonwoven binders which contain formaldehyde, primarily from the self-crosslinking co-monomer N-methylolacrylamide.

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First Named Inventor	Joel Erwin Goldstein			
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Application Number	/			
Filing Date				
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As a below named invento	or, I hereby declare that:									
My residence, post office address, and citizenship are as stated below next to my name.										
I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint invento r (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled:  Reduced Formaldenyde Nonwoven Binders which Contain Polymerized Units of N-Methylolacrylamide										
the specification of which	(Title	of the Invention)								
is attached hereto	•									
OR was filed on (MM/DI	omm)	as United	States Applicati	on Number or PCT International						
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I acknowledge the duty to di	sclose information which is m	aterial to patentability as de	fined in 37 CFR	1.56.						
I hereby claim foreign priorit certificate, or 365(a) of any America, listed below and har or of any PCT international a	PCT international application	which designated at least	application for r	natent or inventor's cer tificate.						
Prior Foreign Application	Country	Foreign Filing Date (MM/DD/YYYY)	Priority Not Claimed	Certified Copy Attached? YES NO						
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# **DECLARATION** — Utility or Design Patent Application

hereby claim the benefit under 35 U.S.C. 120 of any United States application(s), or 365(c) of any PCT international application ion designating the Jnited States of America, listed below and, insofar as the subject matter of each of the claims of this application is not disc losed in the prior a little States or PCT International application in the manner provided by the first paragraph of 35 U.S.C. 112, I acknowledge the duty to disclose prior application which is material to patentability as defined in 37 CFR 1.56 which became available between the filing date of the prior application and the national or PCT international filing date of this application.										
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As a named inventor, I her	eby appoint the fo	lowing re	gistered practiti	oner(s)	to prosecute this	application a	nd to trai	rsact al	I business in	
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Joel Erwi	n				Golds	stein				
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valid OMB control number.

## **DECLARATION**

### **ADDITIONAL INVENTOR(S)** Supplemental Sheet Page \_\_1 of \_\_1

Name of Additional Joint Inventor, if any:  A petition has been filed for this unsigned inventor										
Given Na	me (first and middle [if any]) Family Name						ne or s	Sumame		
Christian Le	onard				Dar	niels	iels			
Inventor's Signature	Christ L. &	Jan	je D	<u> </u>		**************************************	ž	26 0d 20 Date	œ	
Residence: City	Macungie	State	PA	c	ountry	USA		Citizensh	ip U	SA
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Name of Addition	nal Joint Inventor, if a	ny:			A petitio	n has been filed	d for t	his unsigi	ned in	ventor
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Chung-l	-Ling Mao									
Inventor's Signature	Ching &	Pm	lav	<i>co/26/00</i> Date						
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